REACTIONS OF HEXAFLUOROPROPENE WITH C,N-DIPHENYLNITRONE; A NOVEL SYNTHESIS OF 2-AZETIDINONE

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Little has been reported about the reaction of 1,3-dipoles with hexafluoro-propene (HFP) 1,2. In general, five-membered ring compounds are obtained in the reactions of 1,3-dipoles with alkenes 3. In the course of the study of cycloaddition of 1,3-dipoles with alkenes, we found, however that the reaction of C,N-diphenylnitrone with HFP gave 2-azetidinone instead of a five-membered ring compound. This is a novel example that a four-membered ring compound was obtained from the reaction of a 1,3-dipole and an alkene so far as the authors know 4. This paper describes the elucidation of the structure of resulting product and a possible mechanism to give it.

The reaction of C,N-diphenylnitrone with HFP was carried out in an autoclave at room temperature. From the dark-brown reaction mixture colorless crystals were obtained⁵, mp.;110.0-110.5, which were identified as 1,4-diphenyl-3-fluoro-3-trifluoromethyl-2-azetidinone(1).

The structure of (1) was determined by the spectroscopic methods. ir(KBr): 1762cm^{-1} (C=O), 1192cm^{-1} (C-F). The amide carbonyl band at 1762cm^{-1} strongly suggests a strained ring structure. $^{1}\text{H-nmr}(\text{CDCl}_{3},\text{ppm}(\text{TMS}))$: $5.36(1\text{H},\text{doublet},\text{C}_{4}\text{-H},\text{J}_{\text{H-F}}\text{=}3.4\text{Hz})$, 7.1-7.4(10H,multiplet,phenyl); $^{19}\text{F-nmr}(\text{CDCl}_{3},\text{ppm}(\text{CF}_{3}\text{COOH}))$: $0.0(3\text{F},\text{doublet},\text{CF}_{3},\text{J}_{\text{F-F}}\text{=}11.2\text{Hz}),+103.5(1\text{F},\text{doublet}\text{ of quartet},\text{CF},\text{J}_{\text{H-F}}\text{=}3.4\text{Hz},\text{J}_{\text{F-F}}\text{=}11.2\text{Hz})$; mass(m/e, (rel.int.) (fragmentation) (calcd.)): 309.0763(134) (M) (309.0777), $190.0403(255) \text{ (M-C}_{6}\text{H}_{5}\text{NCO}) (190.0406)$, $119.0379(255) \text{ (C}_{6}\text{H}_{5}\text{NCO}) (119.0371)$, $91.0405(162) \text{ (C}_{6}\text{H}_{5}\text{N}) (91.0422)$, $77.0384(145) \text{ (C}_{6}\text{H}_{5}) (77.0391)$; elemental analysis(obs.(calcd.)): C;61.93(62.14), N;4.48(4.53), H;3.55(3.58), F;24.72 (24.57); UV:: λ max=257nm.

Two fluorine atoms were eliminated in the reaction product. The fact that

Solvent	Reaction Time(hour)	Compound added	Yield(%)
Acetone	40	none	14
	117	H ₂ O (2ml)	16
Ħ	26	Pyridine (2.5 mol.eq.)	22
*	24	Et_3N (2.7 mol. eq.)	trace ^{a)}
CH ₂ Cl ₂	47	none	2.5
Toluene	46	none	15

Table 1 Yields of (1) under Several Reaction Conditions

a) from thin layer chromatography (Rf=0.56, benzene, silica-gel)

the reaction mixture was strongly acidic indicated the generation of hydrofluoride Table 1 shows the yield of (1) under several conditions. On addition of pyridine to the reaction system, pyridine hydrofluoride was obtained and a higher yield of (1) resulted in a shorter reaction time. Triethylamine did not give a good result presumably because it induced the oligomerization of HFP⁶. It was reported that acetone reacted with hexafluoropropene dimer⁷. Therefore it is quite possible that acetone has reacted as a proton donner in the generation of hydrofluoride since dichloromethane gave a much lower yield of (1) when used as the solvent.

In the reaction of 1,3-dipoles with HFP, five-membered ring compounds formed first², and such cycloadducts are relatively difficult to be isolated⁸. Therefore it is quite possible that an isoxazolidine forms first and then it chages to (1) giving off two fluorine atoms, though our attempts to isolate such an intermediate were unsuccessful. Details of the reaction mechanism are under investigation, but it seems quite certain that 2-azetidinone is formed by this ring contraction reaction.

References and footnotes

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- 3) See for examples; R.Huisgen; Angew.Chem.Int.Ed.Engl. <u>2</u> 565 (1963), R.Huisgen ibid <u>2</u> 633 (1963), D.St.C.Black, R.F.Crozier and V.C.Davis; Synthesis 205 (1975)
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